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The number  $6.022 \times 10^{23}$ , called Avogadro's number after the 19th-century chemist Amedeo Avogadro, is the number we use in chemistry to represent macroscopic amounts of atoms and

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molecules. Thus, if we have  $6.022 \times 10^{23}$  Oxygen atoms, we say we have 1 mol of Oxygen atoms.

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8 Review Module / Chapters 1-4 \_\_\_\_ 14. Organic chemistry overlaps with physical chemistry. \_\_\_\_ 15. Applied chemistry is used to attain specific goals. Part C Matching Match each description in Column B to the correct term in Column A. Part D Questions and Problems Answer the following questions in the space provided. 22.

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Chemistry is a very fun subject and being a 10th grader it is important for you to ensure that you understand each topic taught before the final quiz. The quiz below is designed to see just how attentive you were in class this past semester, give it a shot and see just how much you remember so far. All the best!

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